

The Study Of Physico- Chemical And Bacteriological Parameters Of River Salandi And Assessment Of Water Quality Index From Hadagada Dam To Akhandalmani, Bhadrak, Odisha, India

Pratap Kumar Panda^{1*}, Rahas Bihari Panda², Prasant Kumar Dash³

¹Department of Chemistry, A.B. College, Basudevapur, Bhadrak, Odisha, India

²Department of Chemistry, VSSUT, Burla, Odisha, India

³Department of Chemistry, Bhadrak Autonomous College, Bhadrak, Odisha, India

Abstract: The River Salandi, originates in well-known biosphere of Similipala reserve forest of Meghasani hill under Mayurbhanj district and joins with the river Baitarani at Tinitaraf ghat, near Akhandalmani, after running through Hadagada dam, Bidyadharapur Barrage, Agarapada Town and Bhadrak municipality. The river, in its course of flow from Hadagada to Akhandalmani containing 134 Kms. of distance receives both treated and untreated effluents from the mines, industrial effluents from Ferro Alloys Corporation (FACOR), agricultural effluents, urban wastes as well as bio medical wastes. This paper deals with the analysis of Physico-Chemical and bacteriological parameters of water samples, collected from nine different monitoring stations during summer (April & May), rainy(August), post-rainy(October) and winter seasons (December & January) in the year 2015 and 2016 by using standard procedures. The water quality index (WQI) calculated for the year 2015 and jointly for 2015-16 by computing mean values of twelve important parameters of the corresponding years with the help of weighted Arithmetic method reveals that different monitoring stations have different water quality due to contamination of hexavalent Chromium, Iron, Calcium, Fluoride and other organic pollutants. The no station comes under quality 'A' and 'B' and the water of river is unfit for drinking purpose without proper treatment.

Keywords: Standard procedures, WQI, weighted Arithmetic method, Hexavalent Chromium.

I. Introduction

Like fresh air, the fresh water is one among the essential and fundamental gifts of nature, required for maintenance and development of both micro and macro organisms in the world. It is an interesting fact that water covers 71% of total surface of the Earth and only 3% of water is fresh. Out of these 3% of water, 2.5% is stored on Antarctica in the form of ice [7]. Now-a-days, it is the rapid unplanned industrialization, urbanization [48] and other anthropogenic activities [16, 26], responsible for the pollution of water pose a great challenge for the survival and maintenance of flora and fauna. A good quality of water reflects that it is good in physically, chemically as well as bacteriologically. The quality of water changes with the change of season and geographical area [33] as there are several anthropogenic activities such as agricultural, domestic and socio-cultural, which contribute to the change of water quality and are changed with the change of season[17, 21, 66].

It can be highlighted that during rainy and post-rainy seasons, the gravity of pollution is more in comparison to summer and winter seasons because the agricultural activities that discharges much more agricultural effluents containing fertilizer and pesticides to the river water and hence it is a paramount factor responsible for the increase of concentration of heavy metals and other pollutants in the river water [1, 21, 54, 57] and enter to the both human and animal body through food chain [70]. In addition to this, anthropogenic activities such as open defecation in the river bed [27], discharge of bio medical wastes and excretion of animal enhance the amount of pathogenic bacteria and protozoa in the river [42, 51]. The water quality of any water body in any season can be studied by measuring important parameters like dissolved oxygen(DO), bio chemical oxygen demand(BOD), P^H, total hardness (TH), Sulphate, Nitrate, Chloride, Phosphate, heavy metals and bacteria.

In the present study, the river Salandi, originated from well-known biosphere of Similipala reserve forest of Meghasani hill under Mayurbhanj district has been taken and it joins with the river Baitarani, near Tinitaraf ghat before it flows down into the Bay of Bengal at Dhamara. A dam has been built across the river Salandi at Hadagada with longitude 86°.18' East and latitude 210°.17' North in Anandapur sub-division of Keonjhar district for irrigation purpose of Keonjhar, Bhadrak and Balasore districts. The river, after passing the Hadagada, receives untreated mining effluents from Nuasahi chromite mining belt while passing through it as there are three chromite mines namely Baula open caste and underground mines, Bangur chromite mines and Nuasahi chromite mines. It is only Bangur chromite mine that discharges one Lakh tones of chromite over per

year and seven lakh tones of over burden are excavated which is the major cause of pollution by total chromium and hexavalent chromium in form of mining discharges and surface run off [18, 43].

Thereafter the river runs through Bidyadharpur barrage, Agarapada town, Industrial belt at Randia(FACOR), Bhadrak Municipality(District Head Quarter) and finally meets with the river Baitarani at Tintaraf Ghat before it merges into the Bay of Bengal at Dhamara. The river, during its course of flow from Hadagada to Akhandalamani travels 134 Kms. of distance and receives both treated and untreated mining discharges, agricultural effluents, industrial wastes, urban wastes, biomedical wastes, forest run off and after all domestic wastes as it is only the natural drainage system in the study area [44]. Therefore, the aforesaid factors are predominantly responsible for the pollution of the river Salandi as reported in daily Odia news paper “Samaj” very often. In this paper water samples from nine different selected monitoring stations have been collected during summer (April & May), Rainy (August), post-rainy (October) and winter (December & January) in the year 2015 and 2016 and have been analysed by using standard procedures established by APHA, 2005 to study physico-Chemical and Bacteriological parameters [3]. Finally mean value for twelve important parameters have been calculated for the year 2015 and 2016 independently and jointly for both the years. The water quality index (WQI) has been calculated by taking mean values of twelve important parameters for the year 2015 independently and jointly for the year 2015-16 for errorless study, comparison and confirmatory conclusion so as to apply it for the welfare being of the society[12, 32].

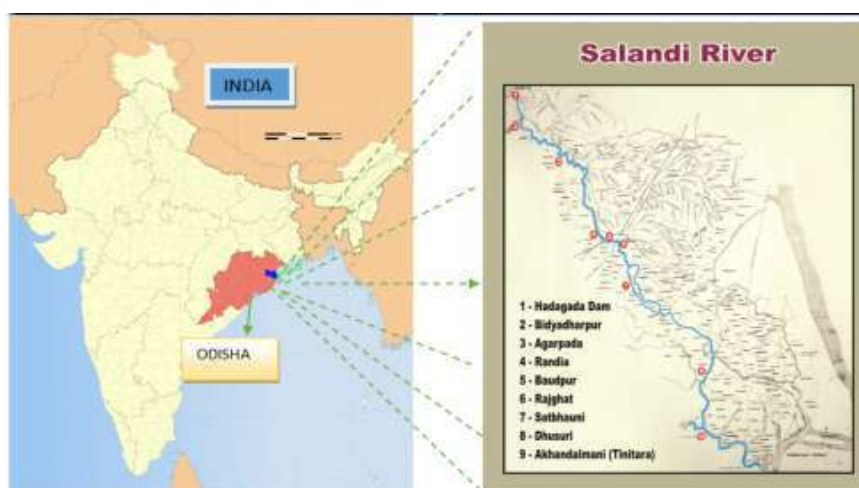
II. Materials and Methods

2.1 Selection of Sampling Stations:

The nine sampling stations have been selected in the bank of river on the basis of gravity of the expected pollutants as well as geography of the river bed to meet the aim and objective. Water samples from nine monitoring stations as located in the map-1 and described in the table-1 have been collected during summer (April & May), rainy (August), post-rainy (October) and winter (December & January) seasons in the year 2015 and 2016 for the analysis and study of water quality of the river Salandi.

Table-1: Location of different Sampling Stations across the river Salandi

SI No.	Name of Stations	Brief Description on Sampling Stations
01	Hadagada Dam	It is 50 KM from Bhadrak town and is a hilly & mining area where the river receives mining, agricultural and forest effluents from Similipal Biosphere.
02	Bidyadharpur	It is nearly 40 KM from Bhadrak town and a barrage is on the river Salandi where it receives mining and agricultural effluents
03	Agarpada	It is 25 KM from Bhadrak town where the river receives agricultural wastes & urban wastes primarily.
04	Randia(FACOR)	At the bank of river Salandi, the village Randia, Ferro Alloys Corporation Industry is established where industrial effluents and agricultural effluents enter into the river.
05	Baudpur	It is 02KM from Bhadrak town where the river receives agricultural effluents.
06	Rajghat	It is situated at the heart of Bhadrak Municipality and nearest to the District head quarter hospital where mainly urban wastes and medical wastes enter into the river.
07	Satbhauni	It is around 20 KM away from Bhadrak town where the river receives mainly agricultural runoff as it is covered with plenty of agricultural lands
08	Dhusuri	It is around 30 KM from Bhadrak town where the river receives mainly agricultural wastes
09	Akhandalmani(Tintaraf Ghat)	It is more than 50 KM from Bhadrak town and is a confluence place of river Salandi & River Baitarani and thereafter the river runs towards Bay of Bengal where the river receives back flow of sea water due to tide and agricultural wastes



Map 1. Location of sampling stations across river Salandi

2.2 Analysis of Physico-Chemical parameters:-

Water samples collected in well cleaned plastic bottles by adding 2 ml of Concentrated HCl in each sample bottles to avoid precipitation of metals have been analysed to study the physico-chemical parameter according to the procedures established by APHA, 2005[3]. TDS and total hardness (TH) have been measured by gravimetric method and complexometric method by using Erio-Chrome black-T as indicator respectively. Further, Calcium concentration has been determined by using EDTA with murexide as indicator. Sulphate has been measured by turbidimetry method. Iron, Chromium and nitrate have been measured with the help of spectrophotometer by using phenanthroline as indicator at 510nm, 540nm and 275nm respectively [44, 51].

2.3 Fluoride, Chloride and Bacteria:-

Fluoride and chloride have been determined by using SPAND reagent and acid zirconium chloride with the help of spectrophotometer at 570nm[41, 45] and titration method respectively. Bacteria have been determined by H₂S kit method [51]. The result of analysis for the month of April, May, August, October, December 2015 and January, April, May, August, October and December 2016 have been placed in the table 2-12 respectively for study and conclusion.

2.4 Calculation of Water Quality Index (WQI):-

The water quality index is a modern method that reflects the quality of any water body under study by means of a single number, calculated by considering combined effect of important physico-chemical parameters. In the present study, weighted Arithmetic method developed by Brown et al in 1972 has been used preferable to calculate the water quality index of the river Salandi [6, 11, 12, 13, 55, 56, 64]. This method includes following steps as explained briefly here under.

i. Calculation of mean value of twelve important parameters

The mean values for twelve parameters independently for the year 2015, 2016 and jointly for the year 2015-16 have been calculated and placed in table 13-15 respectively.

ii. Quality rating scale of each parameter is calculated by using following formula [44, 64]:

$$Q_i = [(V_o - V_i) / (V_s - V_i)] 100$$

Where Q_i = Quality rating of the ith parameter.

V_o = observed value or mean value of the observed values of any parameter.

V_i = ideal value of particular parameter.

V_i = 0 for all parameter except P^H and DO.

V_i for P^H = 7 and for DO V_i = 14.6mg/L.

V_s = Standard permissible value of a particular parameter determined WHO.

iii. Calculation of relative unit weight (w_i)

The relative unit weight of any parameter (w_i) ∝ $\frac{1}{V_s}$

$$w_i = \frac{K}{V_s} \quad \text{Where } K = \text{Proportionality constant}$$

For the sake of simplicity, K is taken as 1

$$\text{iv. Finally water quality index (WQI)} = \frac{\sum w_i Q_i}{\sum w_i}$$

The overall water quality of any water body is assessed by using following gradation table.

Sl No.	WQI Level	Water Quality	Grade
i	0-25	Excellent	A
ii	26-50	Good	B
iii	51-75	Poor	C
iv	76-100	Very Poor	D
v	>100	Unfit for Drinking Purpose	E

In other words

a) $0 < WQI < 100$ *Fit for human use*

b) $0 > WQI > 100$ *Unfit for drinking use [49]*

The WQI for nine monitoring station for the year 2015 and collectively for the year 2015-16 have been given in the table 16 and 17 respectively.

Table 2. Water Analysis Report Of River Salandi In April 2015
Unit - extent pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	6.7	6	89	30	25	3.7	10	4.2	2.8	20	0.40	0.05	0.98	<0.01	7.0	2.5
Bidyadharpur	6.9	5	96	50	45	4.6	8	4.3	2.9	15	0.27	0.20	1.2	0.03	6.7	5.0
Agarapada	7.2	4	73	60	55	4.9	8	4.1	2.9	30	0.38	0.15	1.2	0.07	6.8	3.5
Randia	7.1	4	76	60	55	4.7	12	5.2	3.1	20	0.30	0.17	0.96	0.08	6.7	3.8
Baudpur	7.2	4	73	50	46	3.8	10	5.1	3.1	20	0.30	0.20	0.95	0.05	6.8	2.8
Rajghat	7.1	5	95	30	25	3.6	12	4.9	2.9	30	0.35	0.15	1.0	0.01	6.6	5.8
Satbhauni	6.8	6	95	70	65	4.3	8	4.2	2.8	20	0.30	0.11	1.1	<0.01	6.8	4.6
Dhusuri	6.9	5	92	60	55	4.4	8	4.8	2.6	20	0.30	0.08	0.96	<0.01	6.0	4.2
Akhandalmani	7.3	5	500	450	430	18	8	4.2	2.1	1745	0.43	0.05	0.45	<0.01	7.0	4.2
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.30	0.5	0.6	0.05	6.0	3.0

Table 3. Water Analysis Report Of River Salandi In May, 2015
Unit - except pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	7.0	6	96	80	75	4.5	8	4.1	2.6	25	0.43	0.05	0.93	0.01	7.2	2.6
Bidyadharpur	6.8	5	95	30	25	4.2	7	4.2	2.5	25	0.20	0.04	1.06	0.06	6.8	5.0
Agarapada	7.0	5	86	20	15	4.1	6	4.1	2.4	20	0.39	0.13	0.52	0.07	6.6	3.4
Randia	7.0	4	84	60	55	4.2	6	4.5	2.1	25	0.28	0.16	0.85	0.08	6.4	3.7
Baudpur	6.8	4	82	50	45	4.1	6	4.3	2.1	20	0.25	0.20	0.56	0.04	6.8	2.8
Rajghat	6.5	4	85	60	50	4.5	8	4.1	3.1	15	0.33	0.14	1.1	0.02	6.2	5.7
Satbhauni	6.9	5	85	70	60	4.6	7	4.1	2.9	15	0.21	0.12	1.2	0.01	6.4	4.7
Dhusuri	7.0	5	82	70	60	4.3	7	4.2	2.7	20	0.25	0.06	0.95	0.01	6.6	4.2
Akhandalmani	6.9	5	600	470	450	15	12	5.4	4.2	1750	0.47	0.06	0.42	0.02	6.9	2.0
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 4. WATER ANALYSIS REPORT OF RIVER SALANDI IN AUGUST, 2015
Unit - except pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	7.1	11	105	130	110	8.6	15	5.6	4.1	15	0.36	0.04	0.76	<0.01	7.2	5.0
Bidyadharpur	7.0	10	100	110	100	8.4	13	5.2	3.9	20	0.59	0.30	0.93	0.06	7.0	5.1
Agarapada	6.8	9	98	105	96	6.3	12	5.1	3.4	20	0.34	0.18	0.91	0.05	7.0	5.2
Randia	7.0	10	100	108	98	5.6	12	5.4	3.3	15	0.34	0.17	0.92	0.08	6.4	5.1
Baudpur	7.2	9	96	105	96	5.2	14	5.1	3.5	20	0.32	0.25	0.90	0.06	7.0	5.0
Rajghat	7.0	10	95	100	95	4.5	12	5.2	3.4	20	0.30	0.15	1.1	0.01	6.8	5.3
Satbhauni	7.1	10	98	110	98	4.8	12	5.4	3.8	20	0.30	0.05	0.90	<0.01	7.0	5.0
Dhusuri	7.0	9	100	120	108	5.4	12	5.3	3.5	20	0.30	0.08	0.92	<0.01	7.0	4.9
Akhandalmani	7.2	10	800	480	460	18	15	5.6	4.2	1760	0.45	0.05	0.95	<0.01	7.0	4.4
Bacteria	Positive in all stations.															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 5. WATER ANALYSIS REPORT OF RIVER SALANDI IN OCTOBER, 2015

Unit - extent pH all concentrations are expressed in mg/L

Place	pH	Turbidit	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	7.1	10	108	140	115	8.8	20	5.8	4.8	20	0.30	0.06	0.75	0.01	7.0	5.2
Bidyadharpur	7.3	10	104	155	120	8.4	18	5.4	4.6	25	0.58	0.34	0.91	0.04	7.1	5.3
Agarapada	7.2	9	102	115	100	6.5	15	5.2	4.4	25	0.35	0.20	0.91	0.05	6.8	5.3
Randia	7.1	10	103	110	100	5.8	15	5.6	4.3	18	0.35	0.20	0.92	0.08	6.8	5.5
Boudpur	7.1	8	102	106	98	5.3	16	5.3	4.3	25	0.32	0.22	0.90	0.06	6.8	5.3
Rajghat	7.2	10	100	110	98	4.8	15	5.4	4.2	25	0.30	0.18	1.0	0.03	6.4	5.3
Satbhauni	7.1	10	100	120	100	5.3	16	5.6	4.4	25	0.30	0.05	0.89	0.01	6.7	5.3
Dhusuri	7.1	9	100	126	110	5.6	17	5.6	4.5	25	0.30	0.07	0.91	<0.01	6.7	5.2
Akhandalmani	7.3	11	820	500	470	20	18	5.8	5.4	1760	0.50	0.06	0.94	<0.01	7.1	4.8
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 6. Water Analysis Report of River Salandi in December, 2015

Unit - extent pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	6.9	07	100	56	40	6.2	12	4.8	3.8	25	0.59	0.04	0.25	<0.01	7.2	5.0
Bidyadharpur	7.2	06	98	60	45	6.1	10	4.6	3.6	20	0.76	0.32	0.25	0.02	7.0	5.1
Agarapada	7.2	05	92	64	48	5.6	10	4.5	3.4	15	0.52	0.18	0.28	0.02	6.8	5.1
Randia	7.4	05	96	66	48	5.8	12	4.6	3.2	20	0.48	0.20	0.15	0.08	6.7	5.2
Boudpur	7.2	05	94	65	45	5.5	10	4.5	3.2	20	0.44	0.18	0.15	0.06	7.0	5.1
Rajghat	7.0	06	92	70	56	5.0	12	4.4	3.6	20	0.44	0.18	0.15	0.02	6.5	5.3
Satbhauni	7.1	06	89	75	58	5.5	10	4.5	3.2	20	0.43	0.02	0.10	0.01	6.7	5.2
Dhusuri	7.2	05	90	80	62	5.5	10	4.5	3.5	20	0.43	0.02	0.10	<0.01	6.7	5.2
Akhandalmani	7.3	08	680	315	275	15	15	5.1	4.7	75	4.9	0.05	0.05	<0.01	7.1	4.8
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 7. Water Analysis Report of River Salandi in January, 2016

Unit-except pH all concentration are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	6.9	07	100	60	42	6.2	12	4.8	3.9	30	0.32	0.02	0.24	<0.01	7.1	5.0
Bidyadharpur	7.2	06	97	66	46	6.1	12	4.7	3.7	25	0.54	0.30	0.24	0.01	6.8	5.3
Agarapada	7.1	05	91	64	48	5.5	12	4.6	3.5	20	0.44	0.15	0.28	0.01	6.8	5.1
Randia	7.3	05	95	72	50	5.7	13	4.5	3.4	35	0.51	0.18	0.14	0.07	6.6	5.3
Boudpur	7.2	05	93	70	47	5.5	12	4.5	3.4	25	0.48	0.16	0.14	0.05	6.9	5.1
Rajghat	7.1	05	92	74	58	5.1	13	4.5	3.6	20	0.48	0.15	0.13	0.01	6.5	5.2
Satbhauni	7.3	05	86	76	59	5.5	12	4.7	3.4	20	0.40	0.01	0.10	0.01	6.7	5.1
Dhusuri	7.4	05	86	82	63	5.6	12	4.7	3.5	20	0.40	0.01	0.10	<0.01	6.7	5.1
Akhandalmani	7.3	08	650	320	275	15	15	5.2	4.8	95	3.1	0.03	0.05	<0.01	7.1	4.8
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 8. WATER ANALYSIS REPORT OF RIVER SALANDI IN APRIL, 2016
Unit - except pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	7.2	05	85	30	25	3.5	08	4.1	2.6	25	0.41	0.04	0.96	<0.01	7.2	2.4
Bidyadharpur	7.3	05	96	52	45	4.7	07	4.2	2.8	20	0.28	0.18	1.20	0.02	6.9	4.8
Agarapada	7.1	05	70	58	52	4.7	07	4.1	2.7	25	0.36	0.14	1.20	0.06	6.8	3.4
Randia	7.2	05	75	60	54	4.6	10	5.2	3.0	25	0.30	0.17	0.95	0.08	6.5	4.0
Boudpur	7.1	05	72	52	47	3.7	09	5.0	3.0	20	0.30	0.18	0.94	0.05	6.7	2.7
Rajghat	7.1	05	92	30	25	3.5	10	4.8	2.8	25	0.36	0.15	1.0	0.01	6.5	5.9
Satbhauni	7.1	04	85	65	60	4.1	07	4.0	2.7	20	0.30	0.10	1.0	<0.01	6.7	4.9
Dhusuri	7.1	04	86	58	52	4.2	07	4.6	2.5	20	0.30	0.06	0.95	<0.01	6.0	4.1
Akhandalmani	6.5	07	580	480	456	20	10	4.6	2.4	1800	1.5	0.06	0.46	<0.01	7.0	4.2
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 9. WATER ANALYSIS REPORT OF RIVER SALANDI IN MAY, 2016
Unit - except pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	6.9	5	95	80	75	4.5	08	4.1	2.5	25	0.45	0.40	0.92	0.01	7.1	2.7
Bidyadharpur	6.8	5	94	35	28	4.3	06	4.1	2.5	25	0.23	0.40	1.05	0.06	6.6	5.1
Agarapada	7.0	5	85	25	18	4.2	06	3.9	2.4	20	0.37	0.12	0.54	0.07	6.5	3.5
Randia	6.8	5	85	60	55	4.2	06	4.4	2.1	25	0.26	0.16	0.85	0.08	6.4	3.7
Boudpur	6.8	5	83	50	45	4.1	06	4.2	2.0	20	0.25	0.16	0.58	0.04	6.5	2.9
Rajghat	6.5	4	85	60	50	4.5	07	4.2	3.2	20	0.35	0.15	1.1	0.02	6.1	5.8
Satbhauni	6.9	5	84	60	55	4.5	07	4.1	2.8	15	0.20	0.12	1.1	0.01	6.5	4.7
Dhusuri	7.0	5	84	55	50	4.2	06	4.0	2.7	20	0.22	0.05	0.95	0.01	6.6	4.2
Akhandalmani	6.9	7	580	450	430	15	12	5.5	4.3	1760	0.49	0.06	0.44	0.02	6.8	2.1
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 10. WATER ANALYSIS REPORT OF RIVER SALANDI IN AUGUST, 2016
Unit- except pH all concentration are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	7.2	12	108	135	112	8.5	15	5.5	4.1	15	5.7	0.03	0.75	<0.01	7.1	5.1
Bidyadharpur	7.1	10	104	117	102	8.2	14	5.4	4.0	15	5.4	0.30	0.92	0.06	7.0	5.2
Agarapada	7.0	10	100	110	97	6.2	13	5.3	3.5	20	4.2	0.17	0.91	0.05	7.0	5.2
Randia	7.1	10	102	112	100	5.5	12	5.4	3.3	20	6.07	0.17	0.91	0.08	6.2	5.2
Boudpur	7.2	09	100	108	96	5.2	12	5.2	3.4	20	5.8	0.24	0.90	0.06	6.8	5.1
Rajghat	7.0	10	100	104	95	4.5	12	5.2	3.4	30	4.1	0.13	1.30	0.01	6.6	5.4
Satbhauni	7.1	09	98	110	100	4.8	12	5.4	3.8	25	2.9	0.04	0.90	<0.01	7.1	5.2
Dhusuri	7.1	09	98	120	110	5.5	12	5.4	3.6	20	2.8	0.06	0.90	<0.01	7.0	5.0
Akhandalmani	7.2	12	800	488	462	18.2	15	5.5	4.3	1750	6.5	0.05	0.95	<0.01	7.0	4.5
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 11. WATER ANALYSIS REPORT OF RIVER SALANDI IN OCTOBER, 2016
Unit - except pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	7.2	10	106	140	115	8.5	18	5.6	4.6	20	1.54	0.05	0.72	0.01	7.1	5.1
Bidyadharpur	7.2	10	105	150	118	8.2	20	5.5	4.6	20	1.44	0.32	0.90	0.05	7.1	5.4
Agarapada	7.2	09	104	115	100	6.5	15	5.2	4.4	18	1.42	0.20	0.90	0.04	6.8	5.5
Randia	7.1	10	103	112	102	5.6	15	5.5	4.2	15	1.02	0.22	0.91	0.08	6.7	5.6
Boudpur	7.1	09	102	108	100	5.3	15	5.3	4.2	20	1.02	0.20	0.90	0.06	6.7	5.4
Rajghat	7.1	10	102	110	100	4.6	15	5.5	4.2	25	1.14	0.16	1.1	0.02	6.2	5.4
Satbhauni	7.1	10	100	120	100	5.2	16	5.5	4.4	25	1.12	0.05	0.86	0.01	6.7	5.3
Dhusuri	7.2	10	100	125	110	5.5	16	5.5	4.4	20	1.12	0.05	0.88	<0.01	6.9	5.0
Akhandalmani	7.2	12	810	500	470	20	20	5.8	5.5	1750	1.73	0.05	0.92	<0.01	7.1	4.6
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 12. WATER ANALYSIS REPORT OF RIVER SALANDI IN DECEMBER 2016
Unit - except pH all concentrations are expressed in mg/L

Place	pH	Turbidity	TDS	TH	Ca ²⁺	Mg ²⁺	SO ₄ ²⁻	NO ₃ ⁻	PO ₄ ³⁻	Cl ⁻	Fe	Total Cr	F ⁻	Cr ⁶⁺	DO	BOD
Hadagada	6.8	06	98	55	40	6.1	10	4.6	3.7	25	1.52	0.05	0.75	<0.01	7.0	5.1
Bidyadharpur	7.1	06	98	60	45	6.1	10	4.5	3.6	25	1.42	0.32	0.84	0.02	7.0	5.3
Agarapada	7.2	05	91	62	46	5.5	10	4.5	3.2	20	1.40	0.18	0.82	0.01	6.8	5.1
Randia	7.3	05	95	65	48	5.8	12	4.6	3.2	20	1.02	0.22	0.86	0.08	6.6	5.3
Boudpur	7.2	05	94	65	45	5.5	10	4.5	3.1	15	1.02	0.18	0.88	0.06	7.0	5.1
Rajghat	6.9	06	93	70	55	5.1	12	4.5	3.5	25	1.12	0.16	1.2	0.01	6.5	5.3
Satbhauni	7.1	06	88	75	58	5.5	08	4.3	3.2	20	1.10	0.02	0.85	0.01	6.6	5.2
Dhusuri	7.2	05	90	78	60	5.5	08	4.3	3.3	20	1.10	0.02	0.86	<0.01	6.7	5.2
Akhandalmani	7.3	08	682	318	275	18	15	5.2	4.5	1750	1.75	0.05	0.91	<0.01	7.1	4.5
Bacteria	Positive in all stations															
Standard value of Drinking Water IS-10500	6.5-8.5		500	300	75	45	150	45	5.0	250	0.3	0.5	0.6	0.05	6.0	3.0

Table 13. MEAN Values for twelve important parameters, 2015

Name of Parameters	Hadagada	Bidyadharpur	Agarapada	Randia	Boudpur	Rajghat	Satbhauni	Dhusuri	Akhandalmani
pH	6.96	7.04	7.08	7.12	7.10	6.96	7.0	7.04	7.2
TDS	99.6	98.6	90.2	91.8	89.4	93.4	93.4	92.8	680
TH	87.2	81	72.8	80.8	75.2	74	89	91.2	443
SO ₄ ²⁻	13	11.2	10.2	11.4	11.2	11.8	10.6	10.8	13.6
NO ₃ ⁻	4.9	4.74	4.6	5.06	4.86	4.8	4.76	4.88	5.22
PO ₄ ³⁻	3.62	3.5	3.3	3.2	3.24	3.44	3.42	3.36	4.12
Cl ⁻	21	21	22	19.6	21	22	20	21	1418
Fe	0.416	0.48	0.396	0.35	0.326	0.344	0.308	0.316	1.35
F ⁻	0.734	0.87	0.764	0.76	0.692	0.87	0.838	0.772	0.562
Cr ⁶⁺	0.0094	0.042	0.052	0.08	0.054	0.018	0.0096	0.0092	0.0112
DO	7.12	6.92	6.8	6.6	6.88	6.5	6.72	6.6	7.02
BOD	4.06	5.1	4.5	4.66	4.2	5.48	4.96	4.74	4.04

Table 14. MEAN Values for twelve important parameters, 2016

Name of Parameters	Hadagada	Bidyadharpur	Agarapada	Randia	Baudpur	Rajghat	Satbhauni	Dhusuri	Akhandalmani
pH	7.06	7.1	7.1	7.10	7.08	6.92	7.06	7.12	7.02
TDS	98.4	99.4	90.0	92.0	90.2	94.4	91.0	91.6	690.0
TH	88.0	82.5	74.0	81.8	76.6	74.8	86.0	87.2	447.2
SO ₄ ²⁻	11.8	11.4	10.2	11.0	10.4	10.6	10.0	9.8	14.4
NO ₃ ⁻	4.78	4.74	4.6	5.02	4.84	4.84	4.66	4.76	5.12
PO ₄ ³⁻	3.5	3.5	3.24	3.16	3.14	3.42	3.38	3.3	4.2
Cl ⁻	22.0	21.0	20.6	21.0	19.0	25.0	21.0	21.0	1762.0
Fe	1.924	1.754	1.55	1.734	1.678	1.414	1.124	1.108	2.394
F ⁻	0.82	0.982	0.874	0.896	0.84	1.14	0.942	0.908	0.736
Cr ⁶⁺	0.0094	0.042	0.046	0.08	0.054	0.014	0.0096	0.0092	0.0112
DO	7.1	6.92	6.78	6.48	6.74	6.38	6.72	6.64	7.0
BOD	4.08	5.16	4.54	4.76	4.24	5.56	5.06	4.7	3.98

Table 15. MEAN Values for twelve important parameters for 2015-16

Name of Parameters	Hadagada	Bidyadharpur	Agarapada	Randia	Baudpur	Rajghat	Satbhauni	Dhusuri	Akhandalmani
pH	6.972	7.081	7.09	7.127	7.1	6.954	7.054	7.109	7.127
TDS	99.09	98.81	90.18	92.18	92.09	93.72	91.63	91.63	684.54
TH	85.09	80.45	72.54	80.45	75.36	74.36	86.45	88.54	434.27
SO ₄ ²⁻	12.36	11.36	10.36	11.36	10.90	11.63	10.45	10.45	10.09
NO ₃ ⁻	4.83	4.73	4.6	5.0	4.81	4.79	4.7	4.8	5.26
PO ₄ ³⁻	3.59	3.51	3.29	3.2	3.2	3.44	3.40	3.34	4.21
Cl ⁻	22.27	21.36	21.18	21.63	20.45	23.18	20.45	20.45	1454.09
Fe	1.09	1.06	0.92	0.99	0.95	0.84	0.68	0.68	1.98
F ⁻	0.72	0.86	0.77	0.76	0.71	0.93	0.81	0.77	0.59
Cr ⁶⁺	0.0093	0.039	0.045	0.079	0.053	0.015	0.009	0.009	0.011
DO	7.06	6.90	6.79	6.54	6.81	6.61	6.61	6.68	7.08
BOD	4.15	5.15	4.57	4.76	4.3	5.49	5.02	4.76	4.08

Table 16. Water Quality of river Salandi for nine monitoring Stations, 2015

SI No.	Name of Stations	WQI	Water Quality	Grade
1	Hadagada	43.1737	Good	B
2	Bidyadharpur	98.3067	Very Poor	C
3	Agarpada	108.7467	Very Poor & unfit for Drinking purposes	E
4	Randia	147.0829	Very Poor & unfit for Drinking purposes	E
5	Baudpur	108.0049	Very Poor & unfit for Drinking purposes	E
6	Rajghat	55.4871	Class 'C' River water	C
7	Satbhauni	44.3559	Good	B
8	Dhusuri	39.2897	Good	B
9	Akhandalamani(Tintarf Ghat)	83.8872	Very Poor & unfit for Drinking purposes	E

Table 17. Water Quality of river Salandi for nine monitoring Stations, 2015-16

SI No.	Name of Stations	WQI	Water Quality	Grade
1	Hadagada	7.8753	Poor quality	C
2	Bidyadharpur	118.502	Very Poor & unfit for Drinking purposes	E
3	Agarpada	120.67	Very Poor & unfit for Drinking purposes	E
4	Randia	176.13	Very Poor & unfit for Drinking purposes	E
5	Baudpur	133.468	Very Poor & unfit for Drinking purposes	E
6	Rajghat	72.81	Poor quality	C
7	Satbhauni	57.059	Poor quality	C
8	Dhusuri	54.62	Poor quality	C
9	Akhandalamani(Tintarf Ghat)	107.71	Very Poor & unfit for Drinking purposes	E

III.Result and Discussion

3.1 WQI:

The water quality index (WQI) of the water body under investigation calculated independently for the year 2015 and jointly for both the years 2015-16 by taking mean value of twelve important parameters has been reflected in the table 17 and 18 respectively for study and conclusion. The WQI values for the year 2015 are 43.1737, 98.3067, 108.7467, 147.0829, 108.0049, 55.4871, 44.3559, 39.2897, 83.88718 and jointly for the year 2015-16 are 71.8753, 118.502, 120.67, 176.13, 133.468, 72.81, 57.059, 54.62 and 107.71 in the monitoring station Hadagada, Bidyadharpur, Agarapada, Randia, Baudpur, Rajghat, Satbhauni, Dhusuri, Akhandalmami respectively. It is construed after study and analysis of data that the WQI is not same for all stations and no station comes under class A, B and D water quality [12, 27, 32, 50]. Further it is concluded after carefully examination and interpretation of two years of result that water quality of river Salandi in the study area is under constant deterioration during the year 2016 in comparison to 2015. Besides, it is observed from the WQI values jointly for the year 2015-16 that, quality of water is of "Grade-C" at Hadagada and comes under "Class-E" in the mining belt and industrial belt from Bidyadharpur to Baudpur and again restores the quality "C" grade from Rajghat towards the downstream except Akhandalmami. The deterioration of water quality during the year 2016 may be due to the increase of mining activities after the withdrawal of restrictions partially on Odisha Mining Corporation (OMC) and Indian Ferro Metal Alloys Corporation (IMFA). The increase in water quality in the downstream can be attributed due to the dilution and self purification capacity of the river by natural process [29, 38, 39, 42]. The WQI values for the year 2015 and jointly for the year 2015-16 have been reflected in the figure-1 for study and conclusion.

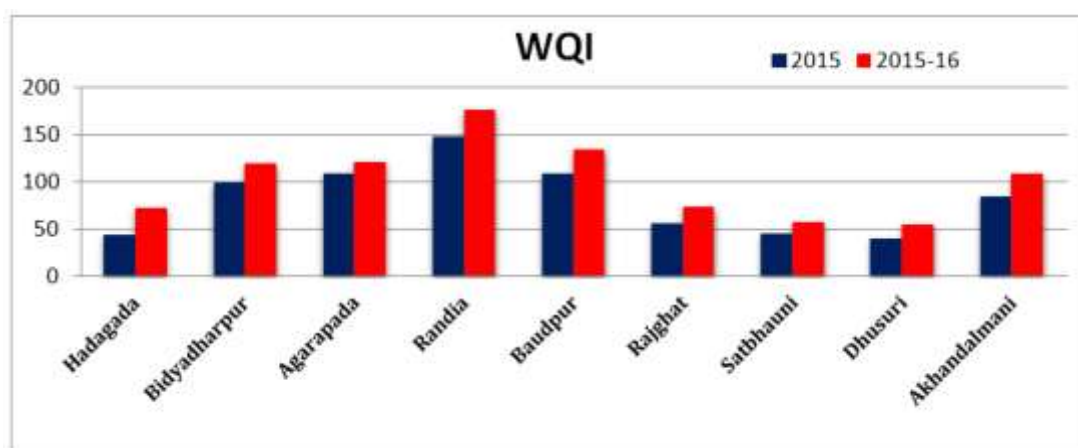


Figure-1: Water quality index for nine monitoring stations during the 2015 and 2015-16

3.2 P^H:-

The P^H is an important parameter required for promotion, maintenance and management of both biotic and abiotic ecological system [25]. The corrosive nature of water is measured by P^H and it is inversely proportional to P^H [40]. The P^H of water body can be changed with the change of season[25, 33], as there are several factors which govern it directly or indirectly are changed with the change of season. The P^H of surface water increases with the increase of photosynthesis by autotrops as they use dissolved CO₂ and release O₂ to the surface of water and low P^H at the bottom of the surface of water is due to the liberation of acids and CO₂ from the decomposition organic maters under low O₂ concentration [10, 21, 25, 28]. The dissolution of CO₂ and Cl₂ in the surface of water forms carbonic acid and hypochlorous acid respectively which fall the P^H [40, 42]. According to Hutchinson (1975), if water body is neither highly alkaline nor highly acidic, the P^H of water body is generally governed by CO₂, CO₃²⁻ and HCO₃⁻ system [63].

The P^H of water samples for nine monitoring stations has been measured from the month of April, 2015 to December, 2016 season wise. It has been observed that, although there is a slight variations but it is within the standard prescribed limit IS-10500(6.5-8.5) [8]. The higher P^H (7.0-7.3) is observed from upstream to downstream in the months from April, 2015 to December, 2016 except May, 2015 and 2016 with certain irregularities that, at Hadagada, Bidyadharpur, Satbhauni and Dhusuri slight lower P^H (6.7-6.9) is observed only during April, 2015 and particular at Hadagada, lower P^H (6.7-6.9) is observed only during winter season for two monitoring years 2015 and 2016. The comparatively lower P^H in the month of May can be attributed due to low flow of water and decomposition of organic materials at high temperature that releases acids and CO₂ [20, 25, 35, 61, 63]. The higher P^H in rainy(August) and post-rainy(October) for entire two years of study can be due to high rate of photosynthesis by autotrops as myxophyceae bloom is observed in the river bed, high flow of rain water and flood that dilutes the pollutants [10, 20, 23, 35, 42, 44, 70].

Besides, slightly lower P^H at Hadagada in comparison to other monitoring stations during winter(December & January) and April for the two years of the study may be due to the decomposition of organic materials and other waste materials thrown off by the large number of picnic parties to the river bed [27, 44] as because Hadagada is a famous picnic spot and attracts large number picnic parties to hold picnic here from the month of December to February in every year, as reported in daily “Samaj” on 12.12.2016. The photo copy of it has been cited in figure-2. The lower P^H at monitoring station Akhandalmani during May (6.9) for two years of study may be due to unseasonal rain fall run off and pre-monsoon rain fall run off in the coastal region as Akhandalmani is very nearer to the Bay of Bengal [43, 44, 51].



Figure-2: the picnic parties are in Hadagada, as reported in daily “Samaj” on 12.12.2016

From the mean value data of two years of study (2015-16) given in figure-3, for nine monitoring stations, it is evident that both Hadagada and Rajghat contain lower P^H that is 6.972 at Hadgad and 6.954 at Rajghat and increasing trend of P^H is observed from Hadagada to Baudpur and it falls at Rajghat and again increases from Satbhauni to Akhandalmani. The lower P^H at Hadagada can be attributed to receiving of forest run off and picnic waste materials and the pollutants are gradually diluted towards downstream and in case of Rajghat, it is due to receiving of biomedical wastes and urban wastes as district headquarter hospital and other private medicals are very close to it and after all the monitoring station Rajghat is situated at the heart of the district headquarter [43, 44, 51]. Further, it may be stated that, after careful comparison of mean value of P^H between two years 2015 and 2016, there is very slightly increasing trend in P^H from Hadagada to Agarapada followed by slightly decreasing P^H from Randia(FACOR) to Rajghat and then increasing trend from Satbhauni to Dhusuri except Akhandalmani during the year 2016. The increasing trend of P^H from Hadagada and its neighboring station Bidyadharpur may be due to high rate of photosynthesis by autotrops [25, 42] and dilution of pollutants due to high flow of rain water and flood during 2016. The decreasing trend in the industrial belt from Randia to Rajghat may be due to mixing of industrial wastes from Ferro Alloys Corporation (FACOR) at Randia [17, 29] and urban wastes and biomedical wastes at Rajghat [44, 51] respectively along with increasing the concentration of chloride in Randia and Rajghat that forms Hypochlorous acids [40]. Further, decreasing trend towards downstream is due to dilution and self stabilization capacity of the river. The comparatively lower P^H at Akhandalmani (7.02) may be due to high unseasonal rain fall run off and pre-monsoon rain fall run off as the monitoring station is nearer to the Bay of Bengal (Sea) [29, 44, 51]. The mean values of P^H for the year 2015 and 2016 have been presented in the figure-3 for better study and conclusion.

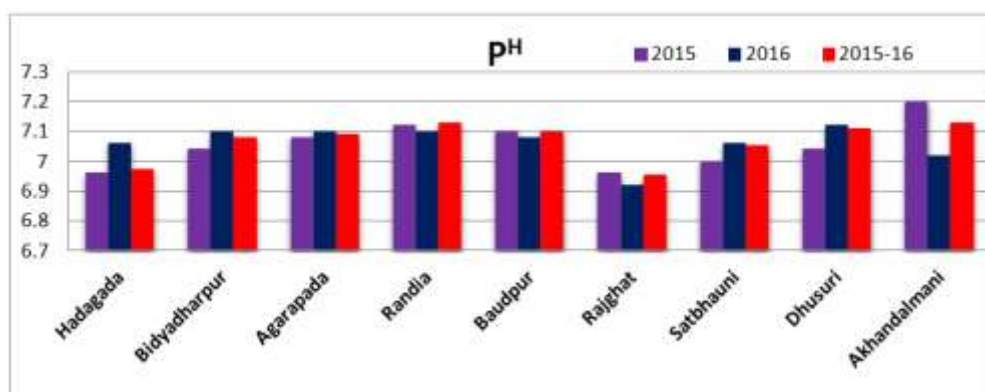


Figure-3: the mean values of P^H for nine monitoring stations during 2015, 2016 and 2015-16

3.3 TDS & Turbidity:-

There is a seasonal variation of TDS and Turbidity in all monitoring stations as it is evident from the analysis result of both the years, 2015 and 2016. It is found from the analysis result that, both TDS and turbidity are higher in all monitoring station from the month of August to October for both the years in comparison to other months. It is due to the entering of high rain fall run off causing erosion of soil, forest run off, agricultural runoff, mining wastes, industrial wastes and domestic wastes to the river water in large scale [25, 27, 51]. Further, it is observed that there is lower value of TDS and turbidity in summer (April & May) in all monitoring stations except Akhandalmani for both the monitoring years (2015 & 2016). It may be due to silt and settling of dissolved salts [36, 40]. The highest value of TDS and turbidity in the monitoring station Akhandalmani for two years of study is due to the back flow sea water from the Bay of Bengal to the river as because the station is nearer to the sea [29, 43]. The mean value of TDS changes 99.09 at Hadagada to 684.54 at Akhandalmani with higher values at Hadagada and in the mining belt Bidyadharpur and its diluted towards downstream in a irregular manner except Akhandalmani. The higher values at Hadagada and its mining belt are due to the mixing of forest run off and mining effluents to the river [50, 51]. Further, after comparison of TDS between the years 2015 and 2016, it is observed that although there is variation of TDS takes place in an irregular way, but it is not significant. The value of TDS for the year 2015 and 2016 has been presented in the figure-4.

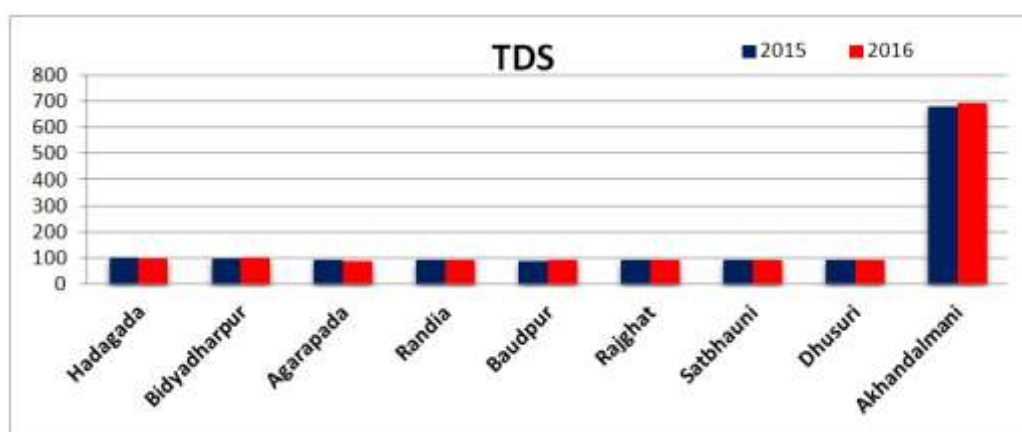


Figure-4: The mean values of TDS for nine monitoring stations during the year 2015 and 2016

3.4 TH, Ca & Mg:-

The seasonal variations are observed in case of TH, Ca and Mg in nine monitoring stations for two years of study, 2015 and 2016. But all that values of TH are within the prescribed limit of IS-10500 (300 mg/L), except Akhandalmani [8]. In the monitoring station Akhandalmani the values are higher than the permissible limit of IS-10500 for all seasons, especially the values are much higher during rainy (August), post-rainy (October) than those of summer and winter seasons. Further it is observed that the mean value of TH changes from 85.09 at Hadagada to 434.27 at Akhandalmani in an irregular manner. The highest value at Akhandalmani is due to the back flow of sea water to river from the Bay of Bengal [29, 43]. The back flow of sea water to the river Salandi with high intensity has been reported in daily "Samaj" on 04.07.2016, 08.09.2016, 12.09.2016, and in daily "Dharitri" on 01.10.2016.

But in case of Mg, all the values are within prescribed limit of IS-10500 (45Mg/L) in nine monitoring stations for two years of study, although there are seasonal variations. All the values of Ca in summer (April and May) are within the prescribed limit of IS-10500 (75Mg/L) and are higher than the permissible limit in rainy (August), post-rainy (October) and in winter season (December & January) for entire period of study. The higher values of Ca for above period in nine monitoring stations can be attributed to in flow of agricultural effluents that contains certain calcium containing fertilizers such as calcium ammonium nitrate ($\text{Ca}(\text{NO}_3)_2 \cdot (\text{NH}_4)_2\text{NO}_3$), basic calcium nitrate ($\text{Ca}(\text{NO}_3)_2 \cdot \text{CaO}$), Calcium superphosphate ($\text{Ca}(\text{H}_2\text{SO}_4)_2 + \text{CaSO}_4 \cdot 2\text{H}_2\text{O}$) used by the farmers in large scale for cultivation purpose during rainy season [35-37, 53, 57, 58]. The another important observation on Ca^{2+} and Mg^{2+} can be highlighted here that, the values of Mg^{2+} are comparatively higher than Ca^{2+} in the month of May than April in monitoring stations Bidyadharpur, Agarapada, Baudapur and Satabhauni for entire period of study. It can be due to the fact that Magnesium hardness increases in the summer than Calcium hardness because of higher CO_2 concentration in the month of May, resulted from the decomposition of organic matter at high temperature forms soluble Magnesium bicarbonate from insoluble Magnesium carbonate. The same thing may not occur in case of Calcium due to its lower solubility as Magnesium compound are generally more soluble than Calcium compounds [52]. Further, it is construed from the comparison of mean values of TH between two years of study that is 2015 and 2016, that

there is significant rise of TH in the year 2016 with higher value at Hadagada (88.0) followed by gradual decreasing trend from upstream to downstream except Akhandalmani. It may be due to excessive rise of concentration of Iron from upstream to downstream in the year 2016. The mean values of TH for both the years have been presented in the figure-5.

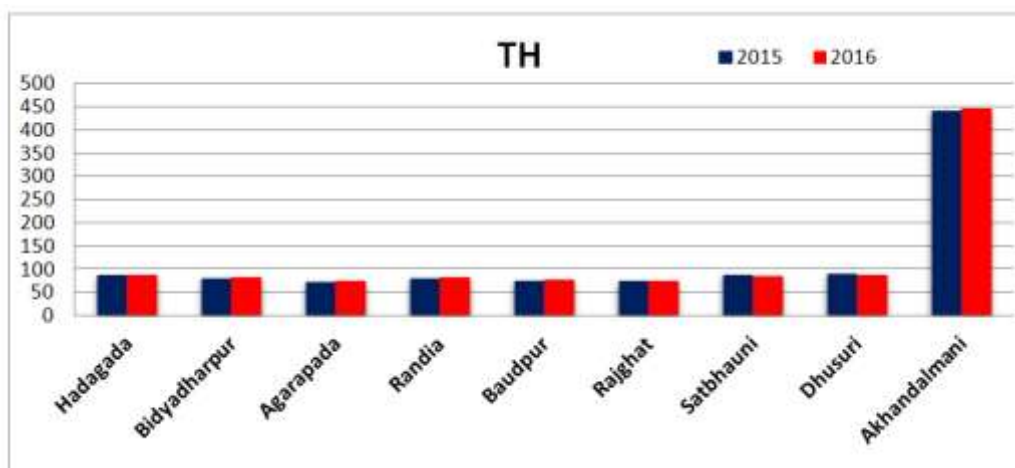


Figure-5: The mean values of TH for nine monitoring stations during the year 2015 and 2016

3.5 NO_3^- , SO_4^{2-} & PO_4^{3-} :-

All the values for above parameters are within the permissible limit of IS-10500 for two years of study, 2015 and 2016. However, the values for above parameters are higher in rainy (August), post-rainy (October) than summer and winter seasons. It is due to the flowing of mining effluents, agricultural effluents, industrial wastes, urban wastes along with forest run off to the river water [17, 25, 27, 35, 51]. The agricultural effluents contribute the problem significantly as pesticides and certain fertilizers used by the farmers in large scale during rainy and post-rainy to promote agricultural productivity contain nitrate, sulphate, superphosphate, which are mixed with the river water as agricultural residues because in ideal conditions, the plant only use 50% of nitrogenous fertilizers applied, 2-20% is lost due to evaporation, 15-25% react with organic compounds of the soil and the rest 2-10% interfere with surface and ground water [2, 36, 53, 57, 58]. The mean values of nitrate, sulphate and phosphate for the year 2015-16 changes from 4.83 at Hadagada to 5.26 at Akhandalmani, 12.36 at Hadagada to 10.09 at Akhandalmani with highest value 12.36 at Hadagada and 3.59 at Hadagada to 4.21 at Akhandalmani in irregular manner respectively. It can be emphasized that all the above parameters show highest value at Akhandalmani due to the back flow of sea water from the sea (Bay of Bengal) to the river [29].

Further, it can be highlighted that the values of nitrate follow decreasing trend from Hadagada (4.83) towards downstream and takes higher value at Randia (5.0) and again decreases towards downstream due to the dilution and self stabilization capacity of the river. The higher values at Randia (5.0) may be due to entry of industrial discharges from Ferro-Chrome plant at Randia (FACOR) to the river water [17, 50]. The value of sulphate is highest at Hadagada (12.36) and decreases towards downstream and again takes high value at Rajghat (11.63) and further decreases towards downstream due to the dilution and self stabilization capacity of the river. The value of phosphate is also higher at Hadagada (3.59) and decreases towards downstream and again rises to 3.44 at Rajghat and further decreases towards downstream due to the effect of same factor. The comparatively higher values of sulphate, nitrate and phosphate at Hadagada may be due to the mixing of forest run off and biological residues from the Similipal reserve forest as Hadagada is nearest to it and picnic wastes left by the large number of picnic parties [27, 43, 44] and again rise up values of phosphate and sulphate at Rajghat is due to the mixing of urban wastes, biomedical wastes [51] and washing residues as district head quarter hospital and other private medicals are very close to it. The laundries use Rajghat is a washing place in large scale [27, 36].

Further, it is evident from the mean values of sulphate, nitrate and phosphate for the year 2015 and 2016 that the changes in concentration of above parameters neither follow definite order nor are they so significant, rather change in values are more or less equal. The mean values of sulphate, nitrate and phosphate for the year 2015 and 2016 have been reflected in the figure-6- 8 respectively for better comparison and study.

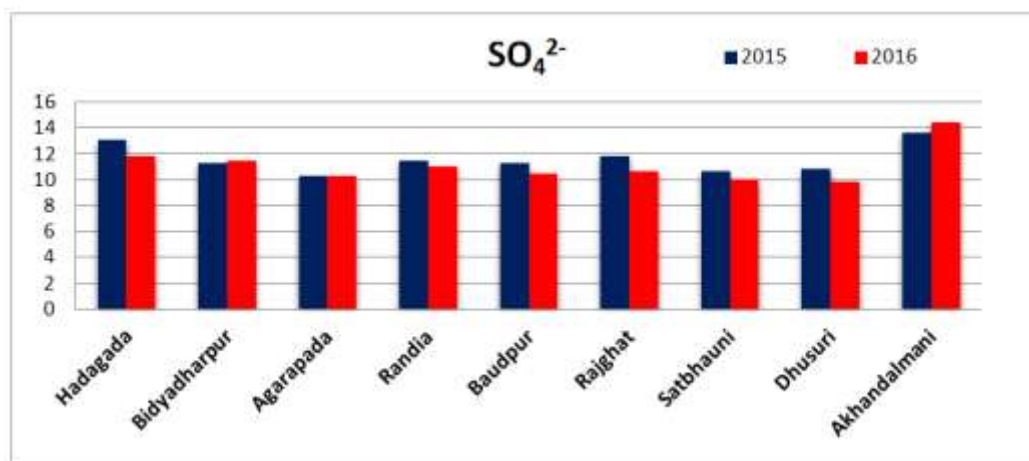


Figure-6: The mean values of SO_4^{2-} for nine monitoring stations during the year 2015 and 2016

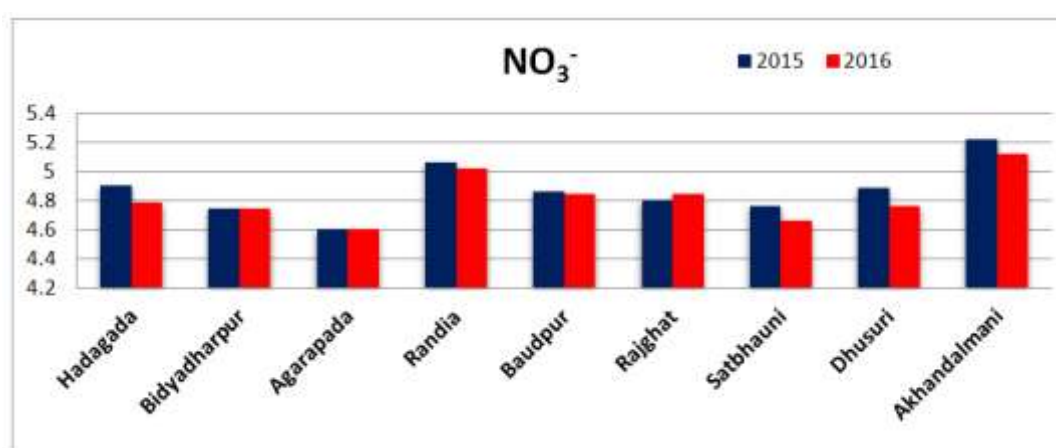


Figure-7: The mean values of NO_3^- for nine monitoring stations during the year 2015 and 2016

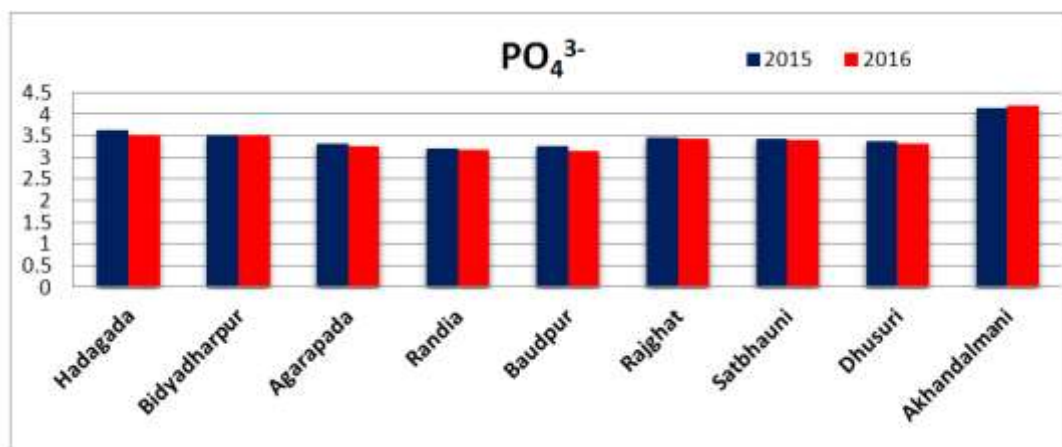


Figure-8: The mean values of PO_4^{3-} for nine monitoring stations during the year 2015 and 2016

3.6 Chloride (Cl):-

The values of chloride in all the monitoring stations are within permissible limit (250 mg/L) except Akhandalmami during two years of study. Although seasonal variations are observed in different monitoring stations but the variation neither follow regular order nor is much significant. From the comparison of mean values of two years of study, it is evident that the value of chloride increase in the year of 2016 very narrowly and highest value is observed at Akhandalmami (1762) followed by Rajghat (25) and Hadagada (22). The overall study for the two years indicates that the mean value of chloride changes from 22.27 at Hadagada to 1454.09 at Akhandalmami with increasing concentration at Hadagada and Rajghat (23.18). The comparatively higher value of chloride in the year 2016 at Hadagada, Randia and Rajghat may be due to increase in

anthropogenic activities such as sewage pollution, industrial activities and septic tank effluents, animal feeds and use of bleaching reagents by the launders [20, 25, 27, 36] and after all mixing of biomedical wastes and urban wastes at Rajghat [51]. The unexpectedly the higher value of chloride at Akhandalmami in both the years of study is due to the back flow of sea water from the sea (Bay of Bengal) to the river [29]. The mean values of chloride for the year 2015 and 2016 have been presented in figure-9 for better comparison.

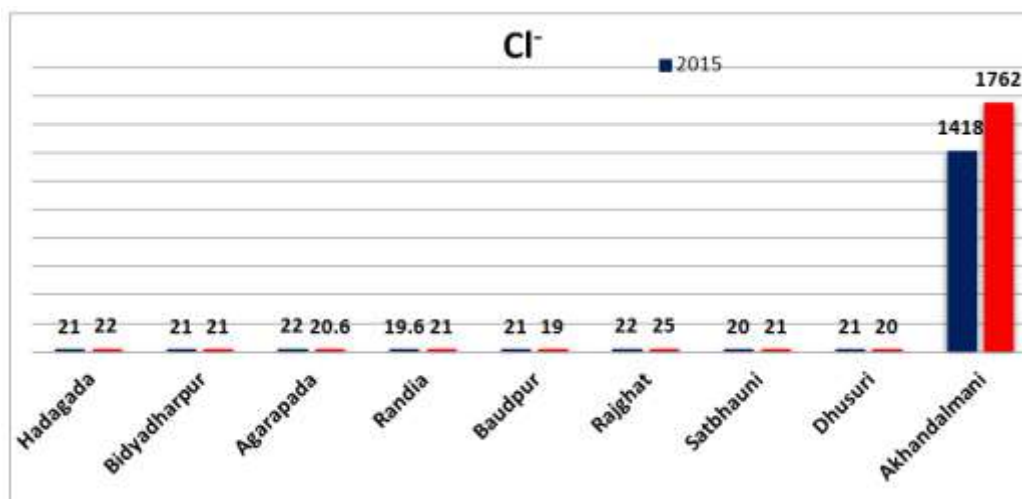


Figure-9: The mean values of Cl⁻ for nine monitoring stations during the year 2015 and 2016

3.7 Fluoride (F⁻):-

The fluoride, responsible for fluorosis is within the permissible range (0.6-1.5mg/L) according to WHO guidelines 2004 and BIS-2003. It is observed that value of fluoride is more in the mining belt (Upstream), Bidyadharpur, Agarapada and gradually decrease towards downstream except Rajghat for entire period of study and these values are higher in summer (April & May) than rainy (August) and post-rainy (October) in all monitoring stations for the above periods except Rajghat (1.1 in August, 1.0 in October, 1.2 in December). Further it is worth mentioning that, these values are higher in the month of May than in April. The mean value of Fluoride calculated jointly for the year 2015-16 changes from 0.72 at Hadagada to 0.59 at Akhandalmami in a decreasing manner from upstream to downstream with the highest value at Rajghat (0.93).

The higher value of Fluoride at Bidyadharpur and its neighboring station Agarapada may be due to the availability of higher soluble compounds of fluorine such as sodium fluoride (NaF), Fluorosilic acid (H₂SiF₆), sparingly soluble compounds of fluorine such as CaF₂ and crylite (Na₃AlF₆) in the soil and rocks [19, 60]. The phosphate fertilizers also contain an average amount of 3.87% of fluoride which can be released to the river as agricultural effluents. Water, when passes through and over the soil and rocks containing fluorine dissolves it and carries to the river [19]. The comparatively lower value of fluoride during rainy (August) and post-rainy (October) is due to the high flow of the rain water and flood that dilutes it [29, 59] and the higher value of fluoride in summer, especially in the month of May can be attributed to very low flow of water and evaporation [27, 57].

Irrespective of nature of seasons and year, the higher value of fluoride at Rajghat and its neighboring station Satbhauni during summer might be the inflow of biomedical wastes, urban wastes to the river as because the district head quarter hospital and other private medicals are at the bank of river [41, 44, 51] and hence fluoride appears at Satbhauni, the neighboring station of Rajghat accordingly and subsequently it is diluted and self-stabilized due to the self-stabilization capacity of the river. It is construed from the mean values that the concentration of fluoride increases significantly in the year 2016 in all monitoring stations with highest value at Rajghat. It might be the dissolution of soil containing soluble compounds of fluoride and highest concentration of fluoride at Rajghat may be due to the mixing of biomedical wastes and municipality wastes in the river water. The mean values of the fluoride for the year 2015 and 2016 have been presented in the figure-10 for better comparison and study.

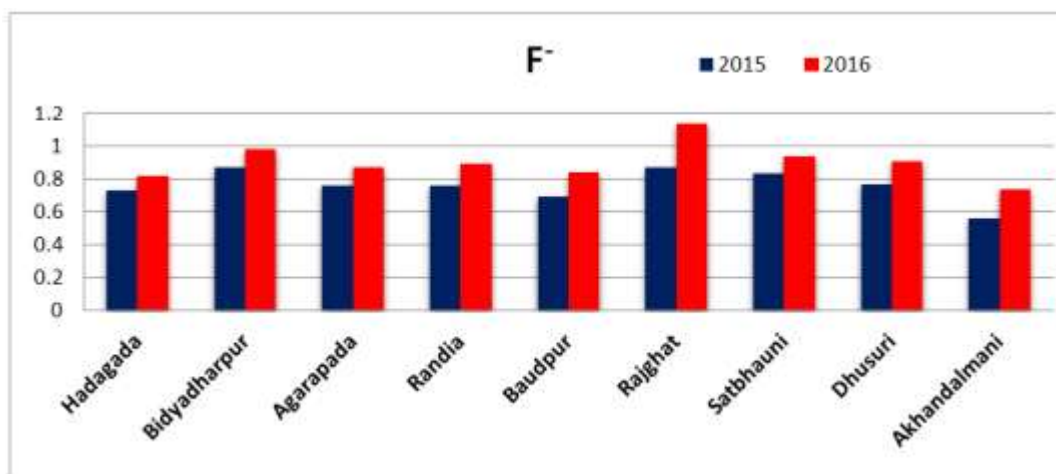


Figure-10: The mean values of F⁻ for nine monitoring stations during the year 2015 and 2016

3.8 Fe (>0.3) & Cr⁶⁺ (>0.05):-

Iron is an important element required for both animals and plants, especially in cellular process. The insoluble Fe³⁺ is reduced to soluble Fe²⁺ in water by bacteria [33] and Fe²⁺ plays a vital role in mammals as a major ingredient of hemoglobin in carrying out oxygen. Chromium is a redox active element that exists in the environment, mainly in Cr³⁺ and Cr⁶⁺. Cr³⁺ is essential element which exhibit both biochemical and physical functions in the human; whereas Cr⁶⁺ is carcinogenic and affects both digestive and respiratory tracts. Cr⁶⁺ has been reported to accumulate mainly in the root of plants where it is reduced to Cr³⁺ in the vacuole.

Iron and hexavalent chromium are found to exceed the permissible or touching the permissible range in certain monitoring stations during summer, rainy, post-rainy and winter season for entire period of study, April, 2015 to December 2016. It is mainly due to receiving mining discharges from Nuasahi Chromite mining belt, industrial discharges from the industrial sector at Randia (FACOR), agricultural wastes and urban wastes from Bhadrak municipality [1, 29, 51].

It is observed that there is excessive concentration of Fe from the month of August to December during the year 2016 than the previous year 2015. Further it is observed that the concentration of Fe is higher in month May (0.43 in 2015 and 0.45 in 2016) than April (0.40 in 2015 & 0.41 in 2016) during 2 years of study at the monitoring stations Hadagada and gradually decreases towards downstream due to dilution. The higher concentration of Fe in the month of May than April at Hadagada can be due to the low flow of water and evaporation [27, 54]. The higher concentration of Fe at Bidyadharpur in the month of August (0.59) and October (0.58) in the year 2015 is due to the missing of mining discharges with the river water. Furthermore, the comparatively higher concentration of Fe in the month of December, 2015 and January, 2016 in comparison to other months of this year from Hadagada to Akhandalmami can be due to interaction of soil containing iron ore with water that rises the concentration of Fe in the river and it happens at the lower part of the river as because water level is very low [54].

The excessive rise in concentration Fe from the month August to December in the year 2016 in comparison year 2015 may be due to the dissolution of soil containing higher amount of Iron ore in the river as mining activities increases due to the lifting of ban partially on the OMC (Odisha Mining Corporation) and IMFA (Indian Ferro Metal Alloys) on the ground of environment, forest and other legal procedures. The another observation can be enlightened that irrespective season and year, the concentration of Fe is always higher in the monitoring station Akhandalmami in comparison to other stations, is due to the back flow of sea water from the sea (Bay of Bengal) to the river water that contains higher concentration of Fe, as it was confirmed from the experiment that in the month of January, water sample was collected from the station Akhandalmami at the time of back flow of sea water due to tide and analysis result shows that the higher concentration of Fe (3.1) at Akhandalmami [29, 51]. Another cause may be cited that Akhandalmami (Tinitaraf Ghat) is the confluence place of the river Salandi and the river Baitarani. In the upstream of Baitarani (Joda & Barbil of Keonjhar district) there are a large number of Iron mines which release high amount of mining discharges containing Iron to the river and hence water and sediments of the river Baitarani contain more concentration of Iron at the confluence place, Tinitaraf Ghat. The higher concentration of Fe has been reported by Dash et al on the river Baitarani in Joda area in 2015 [14, 15].

The mean values of Fe changes from 1.09 at Hadagada to 1.98 at Akhandalmami in a decreasing manner from the mining belt towards downstream due to dilution factor during the two years of study, 2015 and 2016. The mean values of Fe for the year 2015 and 2016 have been presented in the figure-11 for study and comparison.

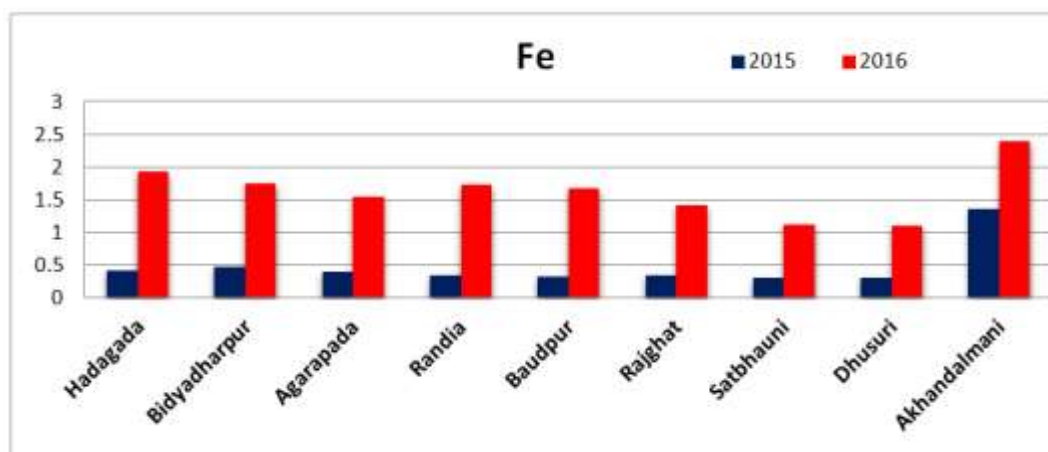


Figure-11: The mean values of Fe for nine monitoring stations during the year 2015 and 2016

As regards hexavalent chromium, its permissible limit is 0.05ppm according to IS-10500 [8]. It is observed that the value of Cr^{6+} is higher in the rainy (August), pos-rainy (October) than the summer and winter seasons for the entire period of study. Further, it is observed that irrespective of the nature of season, the concentration of Cr^{6+} is more in the monitoring station, Randia (0.08) than the permissible value in comparison to other stations. The mean values of hexavalent chromium change from 0.0093 at Hadagada to 0.011 at Akhandalmami with higher value at Randia 0.079. The higher value of Cr^{6+} at Randia(FACOR) is due to the mixing of chromium mixed effluents with the river water from the ferro-chrome plant at Randia [5, 18, 26] and it is decreased towards downstream due to dilution and self stabilization capacity of the river.

Besides, the higher value of Cr^{6+} in the month of August and October for the year 2015 and 2016 may be due to excessive use of chemical fertilizers and pesticides by the farmers during rainy and post-rainy in large scale to increase the agricultural productivity that might contains hexavalent Chromium [1, 5, 24, 29, 53, 54, 57]. At present some chromite mines are not running due to lack of environmental, forest clearances and other legal restrictions imposed by Saha commission headed by ex-justice of the Supreme Court. But earlier report of pollution control Board reveals that they have discharge the effluents of high concentration of Chromium to the river Salandi [18]. Although, at present, some chromite mines are not running, but the mining effluents left by them earlier without proper treatment has resulted open exposure of chromite mixed soil to the atmosphere and pollute the river Salandi with chromium through atmospheric rain precipitations as river Salandi is the only natural drainage system in the study area [5, 9, 30, 47, 54, 68, 69]. Hence more concentration of Cr^{6+} is found from Bidyadharpur to Baudpur in the month of August and October is due to entry of mining discharges, industrial discharges [17], agricultural effluents, atmospheric rain precipitation and geology of river bed and catchment area [29, 67]. The mean value of Cr^{6+} for nine monitoring station for the year 2015 and 2016 have been presented in the figure-12 for better study and conclusion.

It is evident from the figure that the mean value of Cr^{6+} is almost same during the two years of study. It may be due to non-operational of chromite mines as a result of restrictions imposed by Saha Commission.

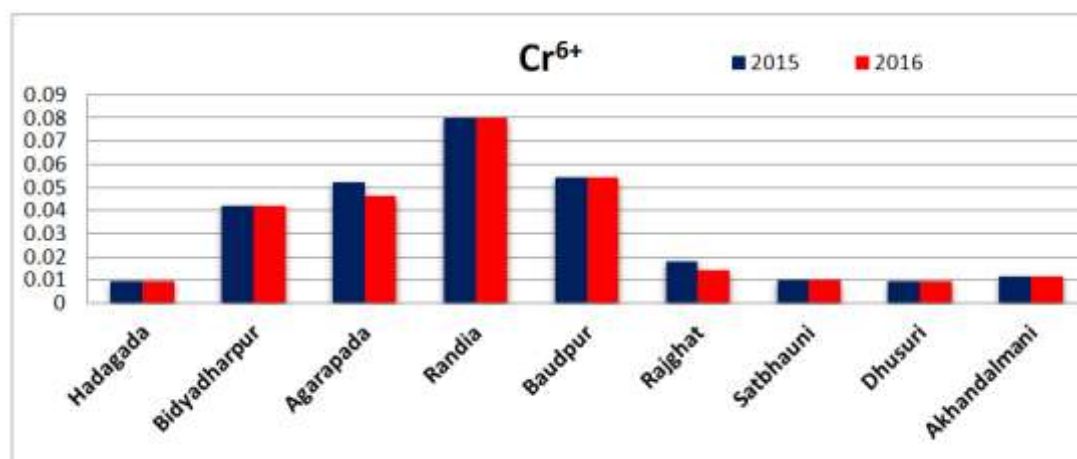


Figure-12: The mean values of Cr^{6+} for nine monitoring stations during the year 2015 and 2016

3.9 Dissolved Oxygen (DO):-

The dissolved oxygen (DO) is a crucial parameter required for survival and maintenance of flora and fauna and to describe the extent of pollution of any water body. Higher the DO, less the pollutants in the water body and vice versa. The major governing factors affecting the concentration of DO at input sources such as dissolution of atmospheric oxygen in water, photosynthesis by autotrops and output sources such as respiration, decomposition of organic matters by the micro organisms and evaporation due to high temperature [25, 28, 35, 42]. Hence concentration of DO in water body increases if input sources are higher than the output sources and minimum value of DO 6mg/L is required for healthy aquatic environment. But it is observed that the DO values have been decreasing from upstream to downstream due to continuous discharge of mining effluents, industrial effluents, agricultural wastes, urban wastes and after all domestic wastes to the river water as dissolved oxygen is being used in redox reaction process to stabilize the pollutants [46]. Further, anthropogenic activities such as open defecation in the river bed throughout the year, picnic [27, 35] and other socio-cultural activities in the river bed during the specific period add fuel to the fire [51].

In the present study, the DO values are constantly affected from Hadagada to Akhandalmani by the aforesaid factors. The mean value of DO for jointly two years of study, 2015-16 changes from 7.06 at Hadagada to 7.08 at Akhandalmani in a gradually decreasing manner with lowest at Randia (6.54), Rajghat (6.61) and Satbhauni (6.61) and further increasing trend is observed in the downstream, from Dhusuri to Akhandalmani with highest at Akhandalmani. The increaseing trend in downstream is due to dilution of pollutants gradually and auto stabilization of pollutants in the due course of flow in the river bed [16]. The highest value at Akhandalmani may be due to high water flow and accumulation of water round the year as it is the confluence place of two rivers that is river Salandi and river Baitarani.

Besides, another observation can be highlighted that the DO values are lower in summer (April & May) than other seasons in all stations for entire period of study except Hadagada. It may be due to the low water flow, evaporation due to high temperature, low rate of photosynthesis by autotrops and after all low rate of dissolution of atmospheric oxygen [15, 20, 27, 31, 34, 35, 42, 62]. The lowest DO at Randia is due to the mixing of industrial effluents from the Ferro-Chrome plant at Randia with the river [1, 25, 50] and the comparatively lower value at Rajghat and its neighboring station Satbhauni may be due to receiving of biomedical wastes and urban wastes as because Rajghat is in the heart of the municipality and district head quarter hospital and other private medicals are very close to it and pollutants appear at Satbhauni through the course of flow of water [51]. Further the higher value of DO at Baudpur (6.87) than its upstream station Randia is due to the dilution of industrial effluents discharged from Ferro-Chrome plant at Randia in due course of water flow. Besides, the higher value of DO in rainy (August) and post-rainy (October) is due to high flow of rain water, flood that dilutes the pollution load, aeration and after all more dissolution of atmospheric oxygen due to low temperature [20, 33, 35, 40, 44, 63].

More ever, it is needless to mention that the DO values are always higher at Hadagada with slight seasonal variation (7.0-7.2) for entire period of study irrespective of the nature of seasons and it can be attributed to higher rate of photosynthesis by autotrops as myxophyceae bloom is observed in the river bed, low pollution load of the river in the upstream, more dissolution of atmospheric oxygen and low rate of eutrophication due to dense forest environment [20, 35, 42, 51, 70]. The mean values of DO for the year 2015 and 2016 have been presented in the figure-13. From the figure it is evident that the DO values in the year 2016 decreases slightly from upstream to downstream except Dhusuri. It may be due to partially running of Odisha Mining Corporation (OMC), Indian Ferro Metal Alloys Corporation (IMFA) and rise of concentration of fluoride. Further it is observed that the concentration of DO remains constant instead of decreasing at downstream Satbhauni followed by increasing at Dhusuri. It may be due to starting of dilution of pollutant from Satbhauni onwards by self stabilization capacity of the river.

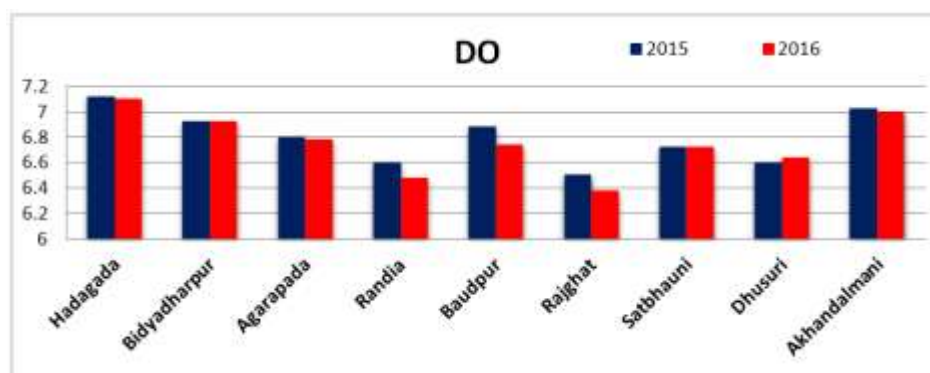


Figure-13: The mean values of DO for nine monitoring stations during the year 2015 and 2016

3.10 Biochemical Oxygen Demand (BOD):-

Like dissolved oxygen, BOD is also an important indicator of water pollution. For any water body, the BOD value, if more than 3 mg/L will be treated as polluted [8]. The higher the BOD values, more the water polluted by the organic pollutants and vice versa. It is observed that the BOD values are either closed to standard permissible limit (3 mg/L) or exceeding the standard pollution limit in some stations and these values are more in rainy (August), post-rainy (October) and winter (December & January) than the summer season (April & May). The mean value of BOD calculated jointly for the year 2015-16 changes from 4.15 at Hadagada to 4.08 at Akhandalmani with the highest value at Rajghat 5.49. Further, from the careful analysis of the result, it is revealed that BOD is higher in mining belt, Bidyadharpur (5.15) and it gradually decreases towards downstream except Rajghat where it takes peak value 5.49.

The higher value of BOD at Bidyadharpur may be due to receiving of mining discharges in large scale as there are three chromite mines namely Baula open cast and underground mines, Bangur chromite mines and Nuasahi chromite mines are in operation. It is the only Bangur chromite mines that discharges one lakh tones chromite ores per year and seven lakh tones of over burdens are excavated [18]. The photo copy of the Boula mines has been cited in the figure 14. The higher value of BOD at Rajghat can be due to mixing of biomedical wastes and urban wastes with river water as district head quarter hospital and other private medicals are at the bank of river [43, 44, 51] and after all it is situated at the heart of municipality containing more than 107369 peoples, together with mixing of washing residues as laundries use this spot for washing purpose in large scale [20, 27, 36]. The higher value during rainy and post-rainy seasons in all monitoring stations can be attributed to high flood and rain water that carries forest run off containing biological residues as study area is very close to the Similipala reserve forest [44, 51], mining discharges, industrial effluents, urban wastes, agricultural waste and after all domestic wastes in large scale [27, 29, 46, 57].



Figure-14: Photo copy of the infrastructure of Boula mines

Further, the higher value of BOD in the month of December and January from upstream to downstream may be due to throwing of picnic wastes [27], washing of motor vehicles in the river [1, 41, 54] as Hadagada is a famous picnic spot and attracts a large number of picnic parties to hold picnic in the river bed from the month of December to February in every year, as reported in daily news paper "Samaj" on 12.12.2016 and it has been cited in the figure-1. Besides, open defecation in the river bed [15, 27, 35], burning and throwing of dead bodies and holding of socio-cultural function in the river bed [44,55] contribute significantly to the problem along with low flow of water and precipitation of contaminated dust through the rain water [9, 30, 54, 68, 69].

The mean values of BOD for the year 2015 and 2016 have been given in the figure-15 for better studies and comparison. It is concluded from the figure that in the year 2016 the BOD value increases from Hadagada to Satabhauni and decreases at Dhusuri and Akhandalmani with highest value at Rajghat (5.56) and Bidyadharpur (5.16). Further, it is observed that the DO values at Hadagada and Akhandalmani are nearly equal and it is highest having lowest BOD value. The highest BOD at Rajghat may be due to mixing of biomedical, urban wastes and washing residues with the river water [20, 27, 43, 44, 51]. The higher value of BOD at Bidyadharpur is due to the mixing of more amounts of mining effluents with the river as result of increase of mining activities during 2016 in comparison to 2015.

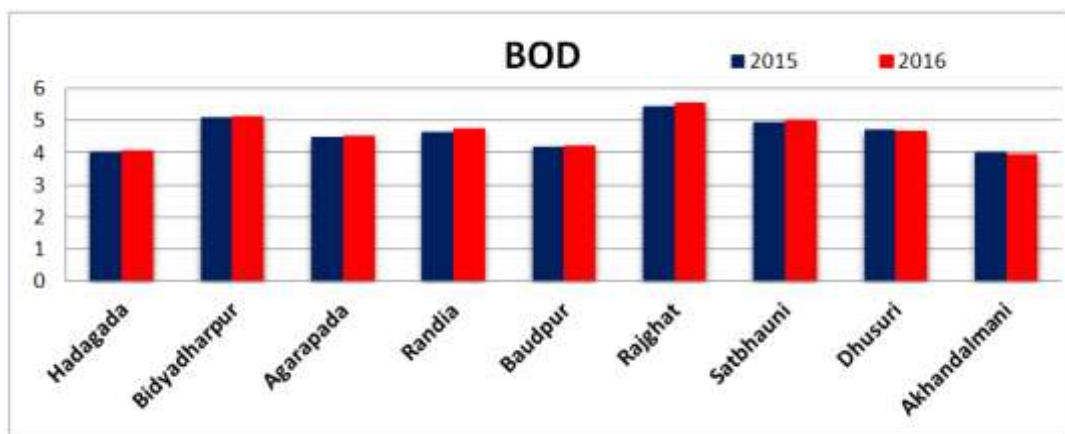


Figure-15: The mean values of BOD for nine monitoring stations during the year 2015 and 2016

3.11 Bacteriological Tests:-

The bacteriological tests have been done in nine monitoring stations during 2015 and 2016 season wise through H₂S kit method. It is found that irrespective of season the bacteria are positive in nine monitoring stations starting from Hadagada to Akhandalmani. It may be due to the mixing of biomedical wastes, open defecation in the river bed, burning and throwing of dead bodies to the river and animal excretion in the river that might contain pathogenic bacteria [15, 35]. Hence the river water is not only polluted physically and chemically but also bacteriologically and unfit for drinking purposes.

IV. Conclusion

The river Salandi, Originated from Meghasani hill of well known biosphere, Similipala reserve forest under Mayurbhanj district, after passing around 134 km distance from Hadagada to Akhandalmani through Hadagad dam, mining belt at Nua sahi Chromite mines, Agarpada town industrial sector at Randia, Bhadrak municipality, finally meets the river Baitarani at Tinitaraf ghat near Akhandalmani, before the Dhamara port, the confluence place with the Bay of Bengal. The river water of salandi, after flowing through forest area from the point of origination of Similipala biosphere to Hadagada dam is preserved in the dam. From Hadagada dam, water is distributed in the control manner to the lower catchment area to meet the requirements of drinking purpose, public use and partly for irrigation of command area land of both Bhadrak and Balasore districts.

The analysis and careful study of results reveals that the river Salandi is contaminated at Hadagada with respect to some parameters such as Calcium, Iron, hexavalent Chromium, Chloride, Fluoride, BOD, and Bacteria. It may be due to receiving of forest decayed residues during the rainy season and accumulation in the Hadagada dam due to stagnancy of water. Besides, anthropogenic activity such as holding of picnic in the river bed in large scale during winter season may be partially responsible for pollution [27]. After Hadagada the river water when passes through Baula Nuasahi, the well known chromite mining belt gets contaminated due to mixing of mining effluents as well as mining run off residues from the mining belt which is confirmed from the analysis reports and findings showing high concentration of Fe, Cr, BOD, and lower DO value [18].

Further the river water receives township effluents from Agarapada town in the downstream, industrial effluents from the industrial sector at Randia (FACOR), municipality and medical wastes from Bhadrak municipality which is indicated from the analysis report that highlights gradually increasing trend of Chromium, Iron, BOD and decreasing trend of DO. It is important to highlight that, during investigation in the year 2015, some mines were not in operation due to restriction imposed by Saha commission's report as well as lack of environmental and forest clearances. But in the year 2016, some mines such as Odisha mining corporation (OMC), Indian Ferro Metal Alloys (IMFA) got partial permission for operation and lifting of ore as well as waste material for which certain mining run off pollutant started entering to the course of flow of the river from upstream to downstream. Here it is observed that the parameters like Iron, Chromium and BOD gradually increased in the river water during 2016 in comparison to the previous year 2015.

Besides, WQI calculated taking mean values for twelve parameters for both the years 2015-16 jointly, reveals that the quality of the water is different at different motoring stations that is poor quality (Class-C) at Hadagada, Rajghat, Satbhauni and Dhusuri and very poor and unfit for drinking purpose (Class-E) in the mining belt (in the upstream), at Bidyadharpur, Agarapada, Randia, Baudapur and at sea shore monitoring station, Akhandalmani [50, 64]. The river water, while flowing upstream to downstream, gets less contaminated due to the dilution and self stabilization capacity of the river [15, 29, 62]. But the WQI values calculated during the previous year, 2015, indicates that water quality is good at Hadagada, Satbhauni, Dhusuri very poor at Bidyadharpur, very poor and unfit for drinking purpose at Agarapada, Randia, Baudapur, and Akhandalmani

[44]. Hence from the comparison of the WQI with the year 2015, it can be concluded that the quality of water is under deterioration due to the increasing of mining activities soon after the partial withdrawal of restriction imposed on mines by Saha commission.

From the above investigations, findings and assessments, it is concluded that the river Salandi is not only contaminated physically, chemically but also bacteriologically and not meeting the river water quality of Class-C water standard as a whole. The bacteriological studies in details have not been done due to the inadequate laboratory facilities. However, in present condition, it is not suitable to use for drinking purpose according to IS-10500 [8] and WHO guidelines [65]. The mining activities, if increases subsequently, will intensify the magnitude of pollution and must pose a great challenge to the survival of flora and fauna, ecosystem and after all on the dwellers.

Hence urgent measures such as disinfection, electro-dialysis and particularly for hexavalent Chromium, reduction with SO₂ in acidic medium followed by lime treatment to precipitate as chromium hydroxide along with approved modern technology should be applied to treat the water carefully by the appropriate authority for the benefit of the people, particularly for Bhadrak and Balasore district.

Acknowledgement

The authors are thankful to the Vice Chancellor, VSSUT, the Principal, Bhadrak Autonomous College, Bhadrak, and Executive Engineer, RW&SS for providing laboratory facilities for carrying out this research work. Further the first author is thankful to Mr. C. K. Sahoo, HOD chemistry, A.B. College, Basudevpur and Mr. L.K. Nayak, Asst. Prof. Dept. of Chemistry, Bhadrak Autonomous college for constant co-operation and encouragement. No fund has been received from any source for this work.

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